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3 **PHYSICOCHEMICAL AND MINERALOGICAL PROPERTIES OF CLAYS FROM FOUR SITES IN**  
4 **SENEGAL FOR THEIR USE IN THE PRODUCTION OF COMPRESSED EARTH BRICKS.**

5  
6 **Abstract**  
7

8 This study aimed to characterize four types of Senegalese clay (Diamniadio, Daga Kholpa, Beer, and Diender) for  
9 producing compressed earth bricks (CEB). The study methodology included geotechnical tests, chemical analyses  
10 (X-ray fluorescence), and mineralogical analyses (X ray diffraction, thermogravimetric analysis, and infrared  
11 spectroscopy).

12 The results show that the materials have varied grain-size distributions: clayey for Diender and Diamniadio (44.5%  
13 and 33.4%, respectively), silty for Beer (53.6%), and sandy for Daga Kholpa (64.9%). The high plasticity indices of  
14 Diamniadio (35.9%) and Diender (60.1%) indicate significant plasticity. X-ray fluorescence revealed a  
15 predominance of SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub> and CaO, whereas XRD identified quartz, kaolinite, calcite, illite, and depending  
16 on the site, montmorillonite, feldspar, dolomite, and anhydrite.

17 These results allow to propose different stabilization methods: the most plastic clays (Diamniadio, Diender) can be  
18 stabilized with lime and/or a lime-cement mixture, while the less plastic clays (Daga Kholpa, Beer) are better suited  
19 for cement stabilization to produce cement-stabilized soil (CEB).

20  
21 **Keywords:** Clay's material, physicochemical characterization, mineralogical analysis, compressed earth bricks.  
22

23 .....  
24 **Introduction:**

25 Over the past few decades, population growth and rapid urbanization in developing countries have led to a high  
26 demand for construction materials. Given the economic and environmental limitations of conventional materials,  
27 such as cement and steel, local materials, particularly clay-based materials, are attracting growing interest in the  
28 construction industry [1-5]. In modern construction, these materials have been the subject of several studies focused  
29 on the production of fired materials, particularly roof tiles and bricks [6]. However, given the environmental and  
30 energy concerns associated with global warming, raw clay materials are becoming increasingly attractive from the  
31 perspective of sustainable construction [6]. They have long been used in construction in various forms, such as  
32 adobe, rammed earth, and cob, and more recently, compressed earth blocks. These represent promising solutions for  
33 building sustainable and affordable housing.

34 In Africa, particularly in Senegal, the use of clay materials is growing rapidly, especially in the production of  
35 compressed earth bricks (CEB) [7-10]. However, some of these properties remain poorly understood, limiting their  
36 widespread use [11]. Several studies conducted in similar contexts have shown that the use of local materials  
37 requires thorough characterization, including particle size, chemical, and mineralogical analyses, to assess their  
38 suitability for producing cement-based composite materials (CEBs) [6,12].

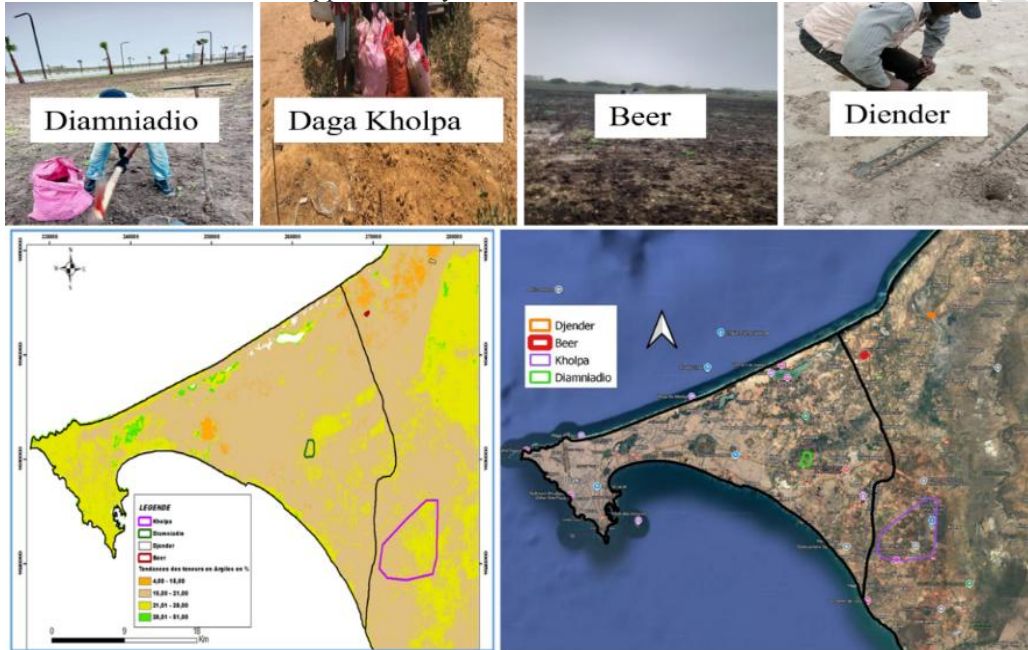
39 The properties of these materials depend heavily on their physicochemical and mineralogical compositions. The  
40 nature and proportion of clay minerals, as well as their particle size and chemical characteristics, directly influence  
41 key parameters such as plasticity, compressibility, texture, and density [6,7,13]. Furthermore, the compressibility  
42 characteristics related to the nature of clay minerals, as determined by oedometer testing, particularly the coefficient  
43 of compression, play a decisive role in these parameters.

44 In this context, the present study aimed to investigate the geotechnical, chemical, and mineralogical properties of  
45 clays from four sites in Senegal to assess their potential for use in the production of compressed earth bricks. The  
46 objective is to better understand the influence of intrinsic parameters, particularly the compressibility parameters of  
47 these clays on the performance of compressed earth bricks (CEB) and to contribute to the development of sustainable  
48 construction solutions adapted to local conditions.

49 **2. Materials and Experimental Methods**

50 **2.1 Materials**

51 The study was conducted on clay soils from four different sites located in the Dakar and Thiès regions of Senegal,  
52 namely Diamniadio, Daga Kholpa, Beer, and Diender (Figure 1), at depths ranging from 0 to 50 cm.  
53 Diamniadio is a municipality located approximately 30 km east of Dakar, Senegal, at 14°42' N and 17°13' W. It  
54 covers an area of 20 km<sup>2</sup> and is expected to accommodate 350,000 residents by 2030–2040. The Daga Kholpa urban  
55 center, covering more than 3,800 hectares, occupies a strategic position between the municipalities of Diass and  
56 Yenne. The site straddles the regions of Dakar and Thiès (at the heart of the Dakar-Thiès-Mbour triangle), 55 km  
57 east of Dakar, 20 km southeast of the Diamniadio Urban Center, and 25 km northwest of Mbour (14° 46' 58" N and  
58 16° 54' 06" W). Beer (14.7828° N, 16.9017° W), located in the Niayes, is a small town in the Thiès region, 70 km  
59 east of Dakar Finally, Diender is a commune located about 30 km from the city of Thiès; it is part of the Keur  
60 Moussa district and covers an area of approximately 116.1 km<sup>2</sup> at 14° 47' 00" N, 17° 03' 00" W.



61  
62 **Fig.1: Sample collection sites.**

63 **2.2 Experimental methods**

64 **2.2.1 Geotechnical identification tests**

65 The organic matter (OM) content of the soil was equal to the ratio of the difference between the initial mass of the  
66 sample (m) and the mass of the sample after reaction with hydrogen peroxide [14].  
67 Particle size analysis was performed using two methods: the coarsest fraction (> 80 μm) was determined by wet  
68 sieving, and the finest fraction (< 80 μm) was determined by sedimentometry in accordance with standard NF P 94-  
69 057 [15].  
70 The particle size distribution was determined in accordance with standard NF EN ISO 17892-4 [16].  
71 The Atterberg limits were determined using a Casagrande apparatus in accordance with standard NF P 94-051 [17].  
72 The quantity, activity, and swelling properties of the clay fraction contained in the material were evaluated using the  
73 methylene blue test in accordance with standard NF P 94-068[18].  
74 The absolute and bulk densities were determined according to the standard NF P 18555 [19].  
75 The oedometer test, which is used to determine the compressibility of a material, was conducted in accordance with  
76 standard XP P 94-090 [20].

77 **2.2.2 Chemical and mineralogical analyses**

78 The chemical composition was determined by extrapolation using regression lines for each element with an x-  
 79 supreme8000 energy-dispersive X-ray fluorescence (XRF-ED) spectrometer. To determine the loss on ignition  
 80 (LOI), the samples were calcined at 1000 °C.  
 81 Thermal Gravimetric analysis (TGA) was performed using a SETSYS Evolution SETARAM-1750 instrument. The  
 82 test involved measuring the mass change of each sample heated to 1000 °C at a rate of 10°C/min in a nitrogen  
 83 atmosphere with a purge flow rate of 20 ml/min.  
 84 The infrared spectra were recorded using an ALPHA II FTIR spectrometer over a range of 400–4000 cm<sup>-1</sup>.  
 85 The mineralogical compositions of the samples were determined using X-ray diffraction. This technique allows for  
 86 the identification of the crystalline minerals that constitute the sample. The instrument used was a BRUCKER D2  
 87 PHASER diffractometer. Determining the proportions of the minerals will be facilitated by semi-quantitative  
 88 analysis of the mineral phases using the relationship proposed by Yvon et al. [21], taken over by Sore [22] and  
 89 Nshimiyimana et al. [6]. Prior to testing, the samples were ground and sieved using an 80 µm sieve.

$$T(a) = \sum Mi Pi(a)$$

90 With

91 T(a): content (in % oxide) of a chemical element « a »

92 Mi: percentage (%) of mineral « i » in the sample under study that contains element « i »

93 Pi(a): proportion of the element « a » in the mineral « i ».

### 94 3. Results and Discussion

#### 95 3.1 Geotechnical properties

##### 96 3.1.1 Organic matter (OM) content

97 *Table 1:* The organic matter content of the samples.

Samples	Dry mass before testing (g)	Dry mass after testing (g)	Organic matter content (%)
Diamniadio	143.13	140.90	1.56
Daga Kholpa	172.95	172.20	0.43
Beer	158.91	150.68	5.18
Diender	172.91	169.25	2.11

98  
 99 Table 1 shows the organic matter content of the samples collected from the four sites. The results indicate that the  
 100 samples from Diamniadio, Daga Kholpa, and Diender have organic matter contents of less than 3%. According to  
 101 Bodian et al. [14], these soils are classified as mineral soils, for which the influence of organic matter on  
 102 geotechnical properties is negligible. In contrast, the beer sample, with an organic matter content of 5.18%, falls into  
 103 the category of mineral soils with organic matter, according to Masi et al. [23]. Thus, except for the Beer sample, the  
 104 negative effects of organic matter on brick-making processes (whether stabilized or not) were negligible.

##### 105 3.1.2 Particle size analysis

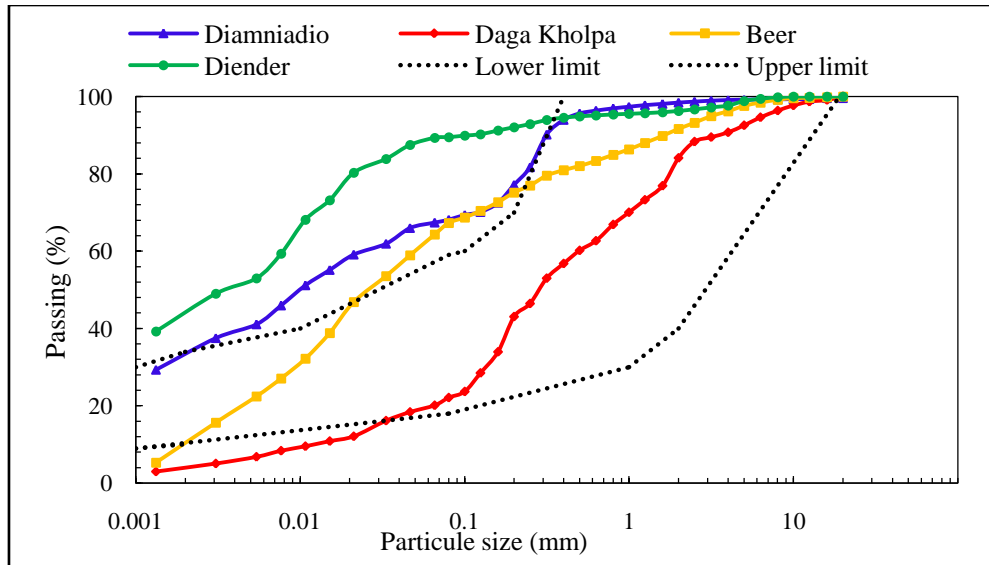


Fig.2: Particle size distributions of the clay soils studied

The particle size distributions of the clays are shown in Figure 2. The following information can be obtained from these curves. Variations in the shape of the curves depending on the location of each sample. The particle size distribution curves for Diamniadio and Diender do not fall perfectly within the limits recommended by CRATerre to produce stabilized bituminous concrete [5]. The Daga Kholpa and Beer curves, on the other hand, extend slightly downward in the case of Daga and toward the middle in the case of Beer. However, these limits primarily serve as guidelines and are not necessarily intended to be strictly adhered to by soil materials [5].

Table 2 explicitly illustrating the distribution of granular particles, which are subdivided into coarse particles (diameter > 2 mm), sand (diameter between 63 and 2 mm), silt (diameter between 2 and 63 μm), and clay (diameter < 2 μm). Overall, the samples from Diamniadio and Diender consisted mainly of clay, those from Beer of silt, and those from Daga Kholpa of sand.

Table 2: Particle size distribution of our samples.

Samples	Clay	Limon	Sand	Gravel	Nature
	< 2μm	2-63 μm	63-2mm	>2mm	
Diamniadio	33.4	34.1	31.1	1.4	Clay-loam
Daga Kholpa	3.7	16.1	64.9	15.3	Sandy-loam
Beer	10.4	53.6	28.2	7.8	Sandy-loam
Diender	44.5	44.9	7.1	3.5	Clay-loam

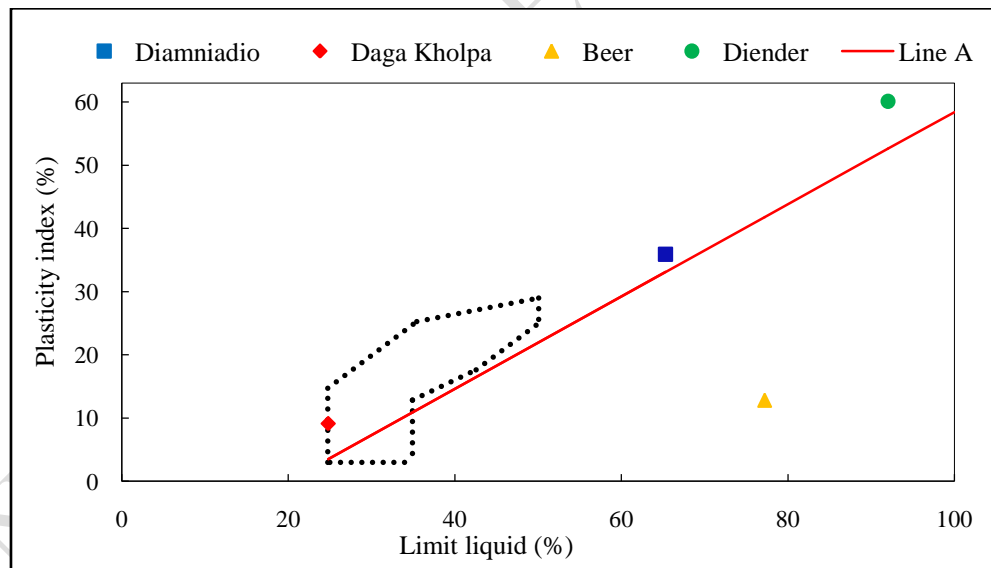
### 3.1.3 Atterberg Limits

**Table 3:** Atterberg limit for the four samples

Samples	$W_L$ (%)	$W_p$ (%)	$I_p$ (%)
Diamniadio	$65.3 \pm 3,8$	$29.4 \pm 1,6$	$35.9 \pm 0,02$
Daga Kholpa	$24.8 \pm 1$	$15.7 \pm 0,4$	$9.1 \pm 0,003$
Beer	$77.2 \pm 3$	$64.5 \pm 12$	$12.8 \pm 0,07$
Diender	$92.0 \pm 4,1$	$31.9 \pm 1$	$60.1 \pm 0,02$

124 The results of the limits for our samples are presented in Table 3. These results clearly distinguish between the  
 125 different samples. Diender (60.1 %) and Diamniadio (35.9 %) exhibited high plasticity indices, characteristic of soils  
 126 ranging from highly plastic to plastic according to the geotechnical classification. In contrast, Daga-Kholpa (9.1 %) and  
 127 Beer (12.8 %) were respectively classified as weakly and moderately plastic. These results are consistent with  
 128 those of the particle size analysis presented in Table 2 and Figure 2. Indeed, Diender and Diamniadio, which have  
 129 the highest clay content ( $< 2 \mu\text{m}$ ) at 44.5% and 33.4 %, respectively, logically exhibit the highest PI values.  
 130 Conversely, Daga Kholpa, dominated by sand (64.9 %), showed a very low IP, whereas Beer, which is  
 131 predominantly loamy (53.6 %), exhibited moderate plasticity. This correlation between the clay fraction and  
 132 plasticity index is commonly observed in the scientific literature [6,14, 24].

133 The relationship between the plasticity index (PI) and liquid limit (LL) is shown in Figure 3.

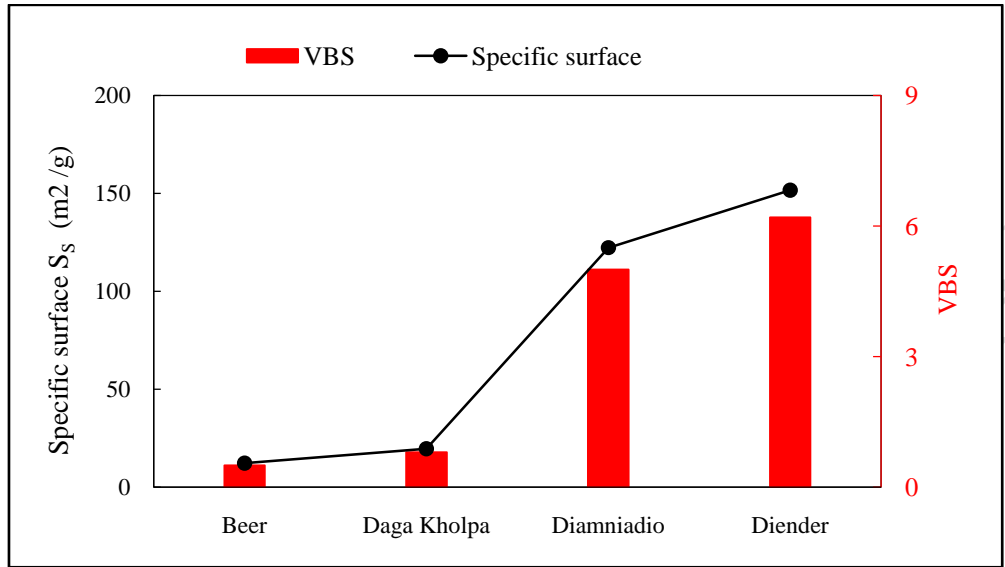


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135 **Fig.3:** Plasticity diagram for the clays studied, compared to the recommendation for CEBs according to XP P 13-  
 136 901 [25].

137 Analysis of the plasticity diagram shows that only the Daga Kholpa sample (PI = 9.1%, LL = 24.8 %) exhibited  
 138 characteristics close to the range recommended for the manufacture of rammed earth bricks [25]. Therefore, it is  
 139 more suitable to produce an unstabilized BTC. The other samples, particularly those from Diender and Diamniadio,  
 140 which fall above Casagrande's Line A exhibit high plasticity index (60.1 % and 35.9 %) associated with high  
 141 liquidity limits (92.0 % and 65.3 %), respectively, indicating a strongly clayey behavior. Therefore, these soils  
 142 require stabilization treatments (addition of mineral binders) to improve their suitability for producing CEB.

143 **3.1.4 Methylene blue value**



144 **Fig.4:** Variation in VBS and specific surface area as a function of sample location.

146 The methylene blue values (VBS) and specific surface areas ( $S_s$ ) of the four samples are presented in Figure 4. The  
 147 results showed that the samples from Diender (VBS = 6.2 and  $SS = 151.70 m^2/g$ ) and Diamniadio (VBS = 5.0 and  
 148  $SS = 122.34 m^2/g$ ) had the highest values for these two indicators. However, Daga Kholpa and Beer exhibited  
 149 significantly lower specific surface areas. These results are consistent with those of the particle size analysis (Table  
 150 2) and Atterberg limits (Table 3). Didier and Diamniadio, which have the highest clay content and IP values, exhibit  
 151 the highest methylene blue consumption and largest specific surface areas. Daga and Beer, characterized by low clay  
 152 content and moderate to low plasticity, showed much lower VBS and  $S_s$  values. The measured specific surface areas  
 153 provide insights into the nature of the clay minerals present in the four samples. According to Santamarina et al.  
 154 [26], the typical specific surface areas are approximately 10–20  $m^2/g$  for kaolinite, 80–150  $m^2/g$  for illite, and 500–  
 155 800  $m^2/g$  for montmorillonite. Thus, the high values obtained for Diender (151.70  $m^2/g$ ) and Diamniadio (122.23  
 156  $m^2/g$ ) suggest the presence of illite or montmorillonite. In contrast, the low specific surface areas of Daga Kholpa  
 157 (19.57  $m^2/g$ ) and Beer (12.23  $m^2/g$ ) may be characteristic of kaolinite-type clays. Taken together, these results can  
 158 guide CEBproduction. The VBS is an indicator of the activity and swelling properties of the clay fraction [18].  
 159 According to CRATerre's recommendation [5] and focusing on the results in this paragraph, it can be seen that:

- 160 • Diender and Diamniadio have high VBS values, indicating an active clay fraction that requires stabilization
- 161 with cement, lime, or a combination of both to control the shrinkage and swelling
- 162 • Daga Kholpa and Beer exhibited low VBS values, suggesting limited clay activity, which makes these soils
- 163 suitable for use with reduced stabilizer addition or even without stabilization for non-structural
- 164 applications.

165 **3.1.5 Absolute and apparent densities**

166 **Table 4:** Variation in absolute and bulk densities as a function of sample location

Samples	Bulk density ( $\rho_{app} kg/m^3$ )	Absolute density ( $\rho_{abs} Kg/m^3$ )
Diamniadio	1099	2366
Daga Kholpa	1299	2604
Beer	662	1082
Diender	1024	2429

167

168 The absolute and bulk densities of the four samples are presented in Table 4. Regarding bulk density, the results  
169 show that Daga Kholpa has  $1299 \text{ kg/m}^3$ , compared to  $1,099 \text{ kg/m}^3$  for Diamniadio,  $1,024 \text{ kg/m}^3$  for Diender, and  
170  $662 \text{ kg/m}^3$  for Beer sample. This difference in bulk density may be due to the organic matter content and the  
171 mineralogical composition of each sample. The particle size distribution shown in Table 2 indicates that the Daga  
172 Kholpa sample contained 80.2 % coarse and sandy particles, compared to 32.5 %, 36 %, and 10.6 % for Diamniadio,  
173 Beer, and Diender, respectively. This indicates that the bricks produced from the Daga Kholpa sample will be denser  
174 than those produced from the other samples.

175 The weight of the solid grains is expressed by the absolute density. We observed the same trend in variation, namely  
176 that the grains from Daga Kholpa weighed more than the other grains. We note  $2604 \text{ kg/m}^3$  compared to  $2366 \text{ kg/m}^3$   
177 for Diamniadio,  $2429 \text{ kg/m}^3$  for Diender, and  $1082 \text{ kg/m}^3$  for Beer.

### 178 3.1.6 Compressibility

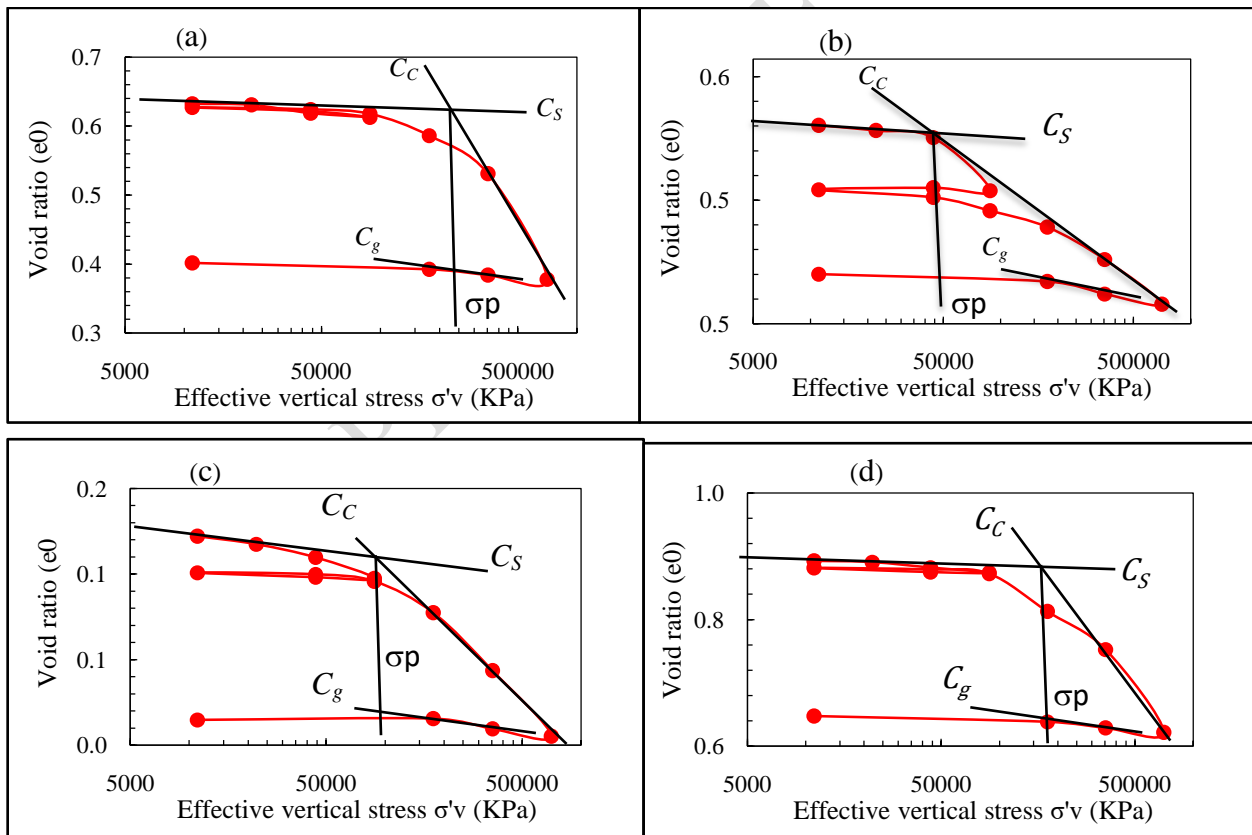
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183



184 **Fig.5:** Compressibility curves for the samples studied: (a) Diamniadio, (b) Daga Kholpa, (c) Beer, and (d) Diender.

185 The compressibility characteristics of the four samples are illustrated by the curves in Figure 5 and are summarized  
186 in Table 5. The main determined parameters were such as the void ratio ( $e_0$ ), compression coefficient ( $C_c$ ),

187 decompression coefficient ( $C_s$ ), consolidation coefficient ( $C_g$ ), reconsolidation stress ( $\sigma_p$ ), vertical stress ( $\sigma_v0$ ), and  
 188 oedometric modulus ( $E_{oed}$ ) (Table 5).  
 189

**Table 5:** The compressibility characteristics of our four samples

Samples	COMPRESSIBILITY CHARACTERISTICS						
	$e_0$	$C_s$	$C_c$	$C_g$	$E_{oed}$	$\sigma_p$ (Pa)	$\sigma_v0$ (Pa)
Diamniadio	0.582	0.004	0.509	0.028	6521413.2	2400000	0.870
Daga Kholpa	0.563	0.010	0.085	0.023	138838.91	50000	0.927
Beer	0.154	0.015	0.127	0.020	4493542	600000	0.697
Diender	0.893	0.006	0.437	0.033	2861218.8	1350000	0.763

190

191 These results show a notable difference between the four samples studied. Diamniadio and Diender have the highest  
 192  $C_c$  values, 0.509 and 0.437, respectively, indicating the high compressibility characteristics of plastic clay soils. In  
 193 contrast, Daga Kholpa ( $C_c = 0.085$ ) and Beer ( $C_c = 0.127$ ) exhibited much lower compression coefficients,  
 194 characteristic of sandy and silty materials with low compressibility, consistent with their low clay content (Table 2)  
 195 and moderate-to-low plasticity (Table 3). The initial porosity index ( $e_0$ ) confirmed this distinction. Diender (0.893)  
 196 and Diamniadio (0.582) had the highest values, reflecting a looser structure typical of clayey materials, whereas  
 197 Beer (0.154) had a very low porosity index, characteristic of a more compact material. The oedometric modulus,  
 198 which represents the stiffness of the material under load, was particularly high for Diamniadio (6.52 MPa) and Beer  
 199 (4.49 MPa), indicating higher resistance to deformation under load. This is advantageous for CEB applications.  
 200 According to the literature [27,28], the values of the compression coefficients ( $C_c$ ) can be linked to the nature of the  
 201 clay minerals present in the samples. Thus, the high  $C_c$  values obtained for Diamniadio and Diender suggest an  
 202 active clay fraction, potentially composed of illite and montmorillonite in Diamniadio and illite and kaolinite in  
 203 Diender. Conversely, the low  $C_c$  values for Daga Kholpa and Beer are typical of kaolinite-dominated materials.

### 204 3.2 Chemical composition

205

**Table 6:** Chemical composition of clays.

Oxides (%)	Samples			
	Diamniadio	Daga Kholpa	Beer	Diender
SiO <sub>2</sub>	58.06	23.50	58.75	27.15
Al <sub>2</sub> O <sub>3</sub>	12.64	6.64	12.83	6.20
Fe <sub>2</sub> O <sub>3</sub>	6.79	7.78	4.4	4.86
TiO <sub>2</sub>	0.76	0.61	0.74	0.55
CaO	6.94	36.43	0.50	23.84
MgO	2.60	0.65	0.83	6.06
SO <sub>3</sub>	0.15	0.10	2.10	3.18
K <sub>2</sub> O	0.42	0.31	1.21	0.77
Na <sub>2</sub> O	0.16	0.13	0.33	2.66
PF	11.44	23.81	17.69	24.71
Total	100	100	100	100
SiO <sub>2</sub> / Al <sub>2</sub> O <sub>3</sub>	4.59	3.53	4.57	4.37
SiO <sub>2</sub> +Al <sub>2</sub> O <sub>3</sub> +Fe <sub>2</sub> O <sub>3</sub>	77.50	37.93	76.54	38.21

206

207 The mass percentages of oxides in different clay materials are presented in Table 6. The chemical compositions of  
 208 the samples, expressed as the mass percentages of their oxides, are presented in Table 6. Chemical analysis showed  
 209 that the four samples were primarily composed of silica (SiO<sub>2</sub>), alumina (Al<sub>2</sub>O<sub>3</sub>), iron oxide (Fe<sub>2</sub>O<sub>3</sub>), and calcium  
 210 oxide (CaO). A note is made regarding the Diender sample; except for potassium oxide (K<sub>2</sub>O) and titanium oxide  
 211 (TiO<sub>2</sub>), all other oxides present were significant.

212 The samples from Diamniadio and Beer were characterized by a high silica content (SiO<sub>2</sub> > 58 %) compared to the  
 213 Daga Kholpa and Diender samples (23 % and 27 %, respectively). This high silica content in Diamniadio and Beer  
 214 samples could indicate a high presence of quartz [14]. The sum of SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, and Fe<sub>2</sub>O<sub>3</sub> for the Diamniadio and  
 215 Beer samples was >70 %, which is consistent with recommendations for stabilized CEBs [6,29].

216 The silica-alumina combination promotes the formation of calcium silicate hydrates (CSH) and calcium aluminate  
 217 hydrates (CAH) during chemical stabilization with cementitious binders. The SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> ratios (4.59 for  
 218 Diamniadio, 3.53 for Daga Kholpa, 4.57 for Beer, and 4.37 for Diender) were similar, indicating a significant  
 219 kaolinite content [14].

220 The high CaO content indicates the presence of either calcite (CaCO<sub>3</sub>) or limestone. This means that the Daga  
 221 Kholpa and Diender samples, which contain significant proportions of CaO (36.45 % and 23.84 %, respectively),  
 222 with a particularly high concentration in the Daga Kholpa sample, where CaO is the main component, are rich in  
 223 calcite.

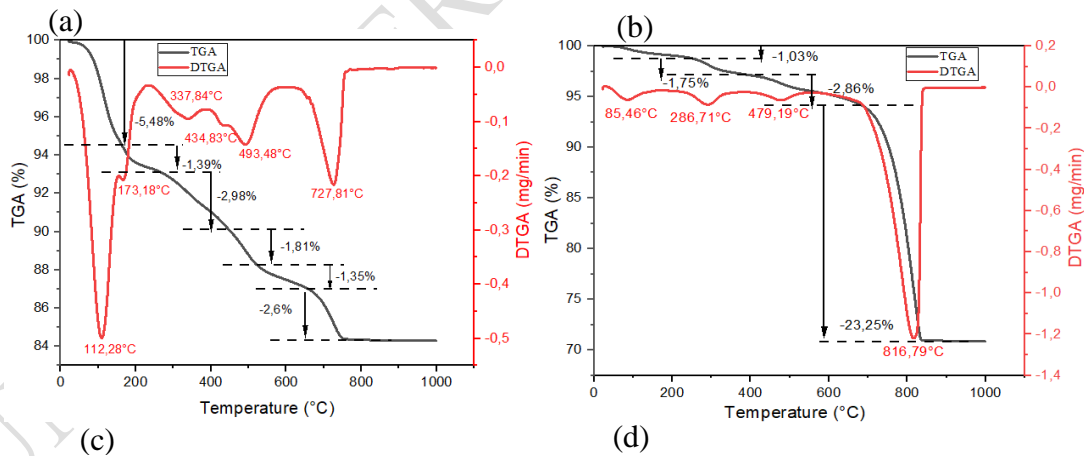
224 The overall composition of the other oxides (TiO<sub>2</sub>, MgO, SO<sub>3</sub>, K<sub>2</sub>O, and Na<sub>2</sub>O) was 4.09 %, 1.8 %, 5.21 %, and  
 225 13.22 %, respectively, indicating that the samples were not pure clays. [30,31]. Given the MgO (2.60–6.06 %) and  
 226 Na<sub>2</sub>O (0.13–2.66 %) contents in some samples, the presence of minerals such as dolomite or feldspar can be  
 227 inferred. The loss on ignition measured at 1000°C was particularly high for Diender (24.71 %) and Daga Kholpa  
 228 (23.81 %). This may be related to the carbonate and/or clay mineral.

229 The chemical composition of the samples determined can guide the choice of stabilizer to produce stabilized clay-  
 230 based concrete using these soil samples. The high proportions of silica and alumina in the Diamniadio and Beer  
 231 samples suggest cement stabilization. In contrast, samples rich in carbonates (Daga Kholpa and Diender) can be  
 232 stabilized with lime.

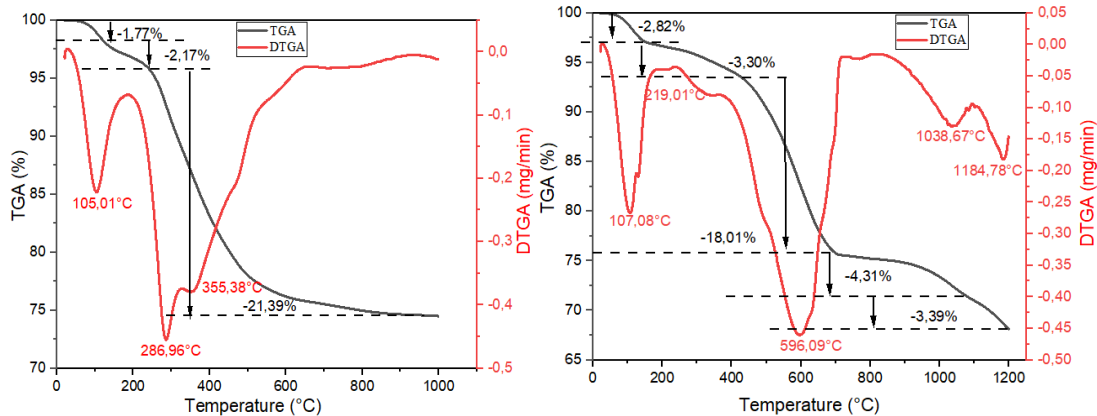
### 233 3.4 Mineralogical composition

### 234 3.3 Thermal Gravimetric Analysis (TGA)

235



236



237 **Fig.6:** Thermogravimetric (TGA) and differential thermogravimetric (DTGA) curves for the clays studied: (a)  
 238 Diamniadio, (b) Daga Kholpa, (c) Beer, and (d) Diender.

239 The TGA/DTGA curves for the samples are shown in Figure 6. The thermograms revealed several endothermic  
 240 peaks, whose positions and intensities varied among the samples. This reflects the mineralogical diversity of the  
 241 samples. In the DTGA curves, four significant endothermic peaks appeared for each sample, except for the beer  
 242 sample.

243 The first peak ( $\approx 100$  °C) corresponds to the removal of free water and intercellular water. The associated mass  
 244 losses were 5.48 % (Diamniadio), 1.03 % (Daga Kholpa), 1.77 % (Beer), and 2.82 % (Diender). Diamniadio stands  
 245 out due to its greater mass loss, indicating a higher affinity for water. This could be related to its larger specific  
 246 surface area and marked plasticity.

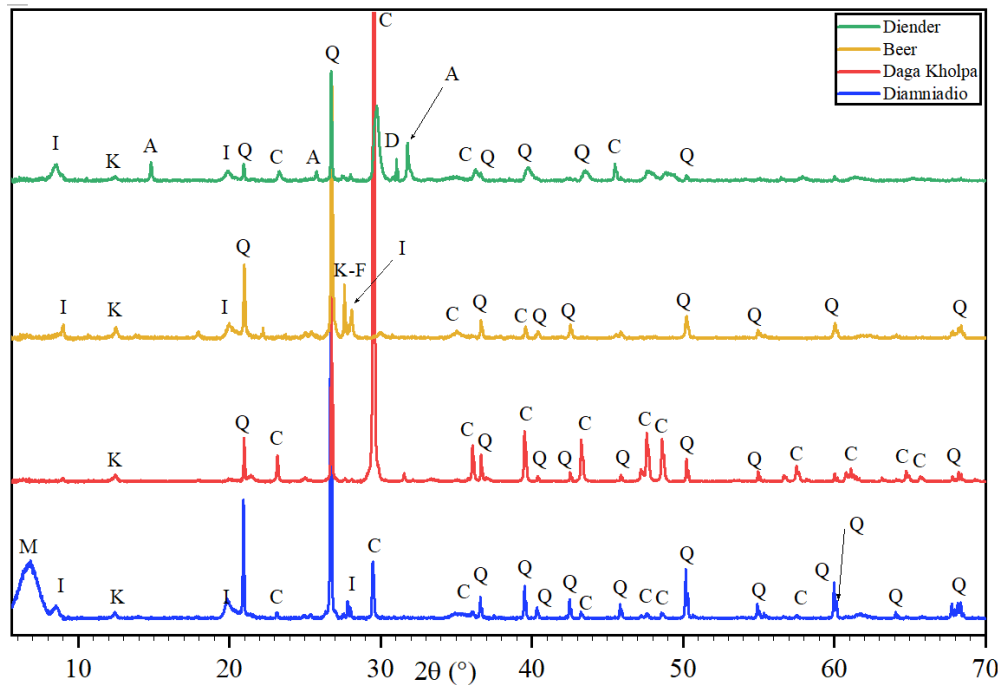
247 The second peak, which appeared at a temperature of approximately 337 °C for the Diamniadio clay, 286 °C for  
 248 Daga Kholpa clay, 286 °C for Beer clay, and 219 °C for Diender clay, was due to the dehydroxylation reaction of  
 249 aluminum and iron hydroxides. At these temperatures, the mass losses were 2.98 %, 1.75 %, 2.17 %, and 2.55 %,   
 250 respectively.

251 The endothermic peaks, which occur in the range of 355 °C to 596 °C for the four samples (434 °C for Diamniadio,  
 252 479 °C for Daga Kholpa, 355 °C for Beer, and 596 °C for Diender), are due to the loss reaction of kaolinite  
 253 hydroxide. Within this temperature range, the Beer clay exhibited the highest mass loss (21.39 %). This may be  
 254 because Beer clay contains the highest proportion of kaolinite.

255 Finally, the peak ( $> 600$  °C), which is absent for Beer, low for Diamniadio (2.6 % loss), but very high for Daga  
 256 Kholpa (23.25 %) and Diender (4.31 %), confirming the abundant presence of calcite and/or dolomite, suggested in  
 257 XRF.

258 Based on these results, it can be concluded that carbonate-rich materials (Daga Kholpa and Diender) may require  
 259 special attention during the drying of CEBs to prevent shrinkage-related cracking [5,32, 33].

### 260 3.4.1 X-ray diffraction (XRD)



**Fig.7:** X-ray diffraction patterns of the four clays studied, with C: Calcite, Q: Quartz, M: Montmorillonite, K: Kaolinite, I: Illite, A: Anhydrite, D: Dolomite, and kF: k-Feldspar.

**Table 7:** Mineralogical composition of the four clays.

Minerals Samples	Kaolinite (%)	Illite (%)	Montmorillonite (%)	Quartz (%)	Calcite (%)	Dolomite (%)	K-feldspath (%)	Anhydride (%)
Diarniadio	11.74	3.56	28.73	31.49	12.39	//	//	//
Daga Kholpa	16.80	//	//	15.69	65.05	//	//	//
Beer	19.17	10.25	//	40.57	0.89	//	7.16	//
Diender	9.34	6.52	//	19.86	27.42	27.88	//	5.41

The mineralogical compositions of the various samples (Figure 7) and the relative proportions of the minerals obtained through semi-quantitative analysis (Table 7) are examined.

XRD analysis revealed that quartz and kaolinite were present in all samples. However, significant differences were observed among the samples.

Diarniadio contains a mixture of quartz, kaolinite, illite, montmorillonite, and calcite; the simultaneous presence of illite and montmorillonite which explains its high plasticity (PI = 35.9 %) and high specific surface area (122.34 m<sup>2</sup>/g).

Daga Kholpa is dominated by calcite (65.05 %) with quartz and kaolinite. This highly carbonate-rich composition is consistent with the high CaO content (36.43 %) and low plasticity.

Beer: primarily quartz, kaolinite, illite, and potassium feldspar. Its low carbonate content (0.89 %) and the presence of feldspar confirm its detrital nature, consistent with its dominant silt fraction.

Diender has a complex mineralogy combining quartz, calcite, dolomite, illite, kaolinite, and anhydrite. The presence of dolomite and anhydrite, as well as the high carbonate content (CaO = 23.84 %), explains its very high plasticity (PI = 60.1 %).

### 3.4.2 Infrared Spectroscopy (IR)

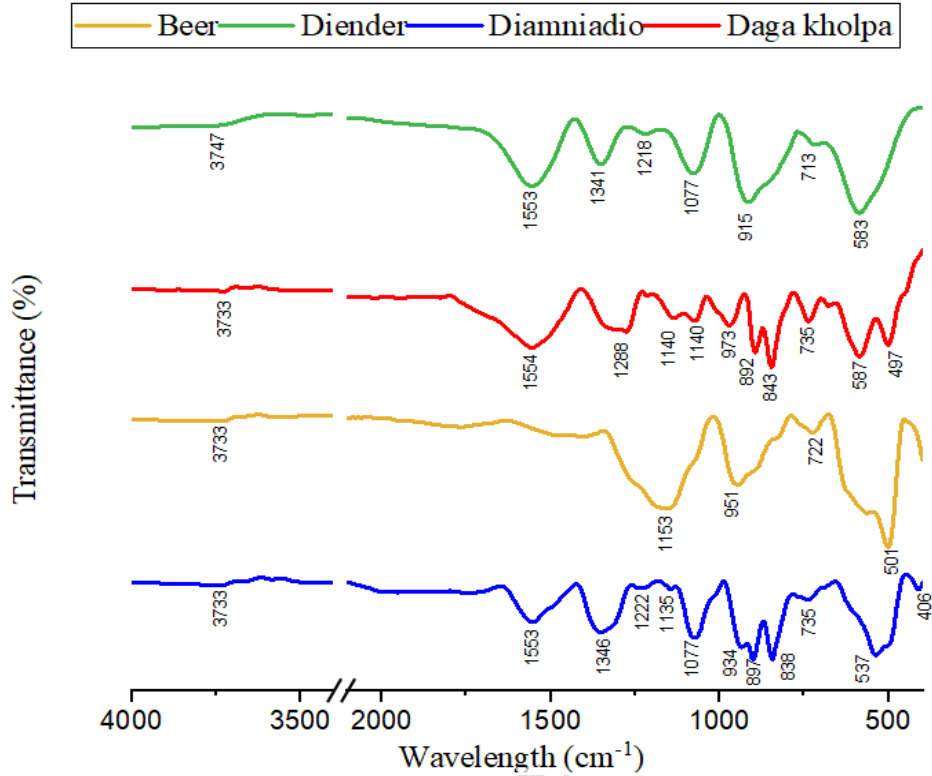


Fig. 8: FTIR infrared spectra of the clays studied in their raw state.

281  
282

283 The infrared spectra of the four clays are shown in Figure 8. Generally, the adsorption bands located around 3500  
284  $\text{cm}^{-1}$  correspond to the structural vibrations of hydroxyl groups characteristic of phyllosilicates [6,14]. The exact  
285 position of these bands and their intensities vary depending on the nature of the molecular bond. They appear closer  
286 for Diamniadio clay at approximately  $3733 \text{ cm}^{-1}$ , for Daga Kholpa clay at approximately  $3733 \text{ cm}^{-1}$ , for Beer clay at  
287 approximately  $3733 \text{ cm}^{-1}$ , and for Diender clay at approximately  $3747 \text{ cm}^{-1}$ . The bands appearing at approximately  
288  $1550 \text{ cm}^{-1}$  correspond to the deformation vibrations of the OH group in the adsorbed water. For all four clays,  
289 we observed significant absorption of radiation between 700 and  $1200 \text{ cm}^{-1}$ ; this indicates the presence of Si-O and  
290 Al-O bonds, corresponding to the valence bond vibration in clays minerals.

#### 291 4. Conclusion

292 This study examined the suitability of clay materials from four sites in Senegal, Diamniadio, Daga Kholpa, Beer,  
293 and Diender, for use as base materials in the production of compressed earth bricks (CEB). The main conclusions  
294 that can be drawn from this study are as follows:

295 Particle size analysis revealed a varied composition: clayey for Diender and Diamniadio (44.5 % and 33.4 %,  
296 respectively), silty for Beer (53.6 %), and sandy for Daga Kholpa (64.9 %). The Atterberg limits show that the high  
297 plasticity indices of Diamniadio (35.9 %) and Diender (60.1%) indicate marked plasticity, conferring strong  
298 cohesion, in contrast to those of Daga Kholpa and Beer (9.1 % and 12.8 %, respectively), which are classified as  
299 non-plastic and moderately plastic. X-ray fluorescence chemical analysis revealed the predominance of  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ ,  
300  $\text{Fe}_2\text{O}_3$ , and  $\text{CaO}$ , whereas  $\text{TiO}_2$ ,  $\text{MgO}$ ,  $\text{SO}_3$ ,  $\text{Na}_2\text{O}$ , and  $\text{K}_2\text{O}$  were present only in small quantities. X-ray diffraction  
301 (XRD) revealed that the main constituents were quartz (31.49, 15.69, 40.57 and 19.86), calcite (12.39, 65.05, 0.89  
302 and 27.42), kaolinite (11.74, 16.80, 19.17 and 9.34) and illite (3.56, 10.25 and 6.52) for Diamniadio, Daga Kholpa,  
303 Beer and Diender respectively. Depending on the sampling site, montmorillonite (28.73) was present at Diamniadio,  
304 K-feldspar (7.16) at Beer, and dolomite (27.88) and anhydrite (5.41) at Diender. These different results allow for the  
305 proposal of differentiated stabilization strategies: the most plastic clays (Diamniadio and Diender) can be stabilized  
306 with lime or a lime-cement mixture, while the less plastic ones (Daga Kholpa and Beer) are better stabilized with  
307 cement to produce CEB.

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