

1 Analysis of The Combined Adsorption-Microfiltration (AMFIL) Process 2 for WWTP Effluent to Process Water 3 4 5

6 **Abstract.** The seaweed processing industry produces Alkali Treated Cotton Chips (ATCC)
7 requires large amounts of process water used for the washing and soaking processes, where process
8 water is generally obtained from groundwater. On the other hand, wastewater from the production
9 process after being treated in Wastewater Treatment Plant (WWTP) and meeting quality standards
10 is discharged into the environment. The effluent of WWTP in large volumes (150 - 200 M³ per
11 day) contains TDS, TSS, turbidity, pH, odor, and organic matter that are relatively small, so it has
12 the potential to be reused as process water. The aims of the research are to investigate the removal
13 of pollutants from the combined adsorption and membrane microfiltration (AMFIL) process on a
14 pilot plant scale from the effluent of WWTP in the seaweed processing industry to process water.
15 The real effluent from the WWTP flows into an adsorption column containing granular activated
16 carbon with varying weights and sizes, then the effluent from the adsorption column flows into a
17 membrane microfiltration with varying weights and sizes membrane. Parameters measured in this
18 study were turbidity, TDS, total hardness, Fe⁺², Mn⁺², and pH. Using an adsorption capacity of
19 0.352 mg.g⁻¹ and the size of the membrane is 5µm can obtain the total removal pollutants of the
20 AMFIL process as turbidity, TDS, total hardness, Fe⁺², Mn⁺², and pH were 86.67,
21 94.00, 95.83, 96.36, 93.33%, and (6.5 - 7.5), respectively.
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23 **Keywords:** Adsorption, Effluent, Membrane Microfiltration, Reuse, Seaweed.
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25 1. Introduction

26 In the seaweed processing industry, producing ATCC (Alkali Treated Cotton Chips) requires a
27 large amount of processed water, whereas for a production capacity of 2 tons of ATCC per day
28 requires process water of ± 150 M³.day⁻¹. Process water is used for the raw material washing
29 process and the seaweed soaking process. Process water is obtained from groundwater and
30 continuous withdrawal of large amounts of groundwater can result in a decrease in environmental
31 quality, in the form of subsidence, landslides, and an imbalance in the distribution of groundwater
32 usage. In addition, excessive groundwater extraction can result in high production costs.

33 Effluent from the WWTP process has a large volume and meets quality standards and is discharged
34 directly into the aquatic environment (river). The characteristics of WWTP effluent are odorless,
35 color (15 - 25 Pt-Co), Turbidity (8 - 14 NTU), pH (6 - 7), TSS (20 - 30 mg. L⁻¹), TDS (4,000 -
36 4,500 mg. L⁻¹), Total hardness (700 - 900 mg. L⁻¹), BOD (30 - 40 mg. L⁻¹), and COD (60 - 80 mg.
37 L⁻¹). By considering the potential and characteristics of WWTP effluent and problems in obtaining
38 process water, the WWTP effluent has the potential to be reused as process water through a
39 combination of adsorption (GAC) and microfiltration (MF) processes which is known as the
40 AMFIL process.

41 Granular activated carbon (GAC) fixed bed adsorption technology was applied to remove
42 carbamates carbaryl, methomyl, and carbofuran from a public water supply. To minimize the
43 effect of clogging and to evaluate adsorbent saturation for carbamates, microfiltration (MF) was
44 previously used to adsorb, and the backwash procedure of the GAC bed was carried out. The MF
45 of the water previously adsorbed in the GAC fixed bed can removal of 100 % of the carbamate's
46 pesticides with an initial concentration of 25 µg.L⁻¹ during the first 48 h of operation (Alves et al.,
47 2020). The combination of adsorption and microfiltration processes has the ability to remove
48 contaminants in water such as hardness, nitrate, and heavy metals (Bakalár et al., 2019). The
49 combination of adsorption and deep-bed filtration proved to be a robust and economic process
50 combination for the removal of micropollutants with economic advantages over membrane
51 filtration (L Pillay & Jacobs, 2007). PAC/Alum/MF achieved 75% to complete removal of total

52 microcontaminants with 4-18 mg. L⁻¹ of a mesoporous PAC and 2 h contact time, with a reliable
53 particle separation (turbidity < 0.03 NTU) and low aluminum residuals(Campinas et al., 2021).
54 In wastewater treatment, membrane technology is called a major technology in the field of water
55 treatment in the 21st century. With the development of membrane technology and the development
56 of other emerging technologies in combination, microfiltration membrane technology is widely
57 used in the treatment of various types of wastewaters such as radioactive wastewater and heavy
58 metal wastewater(Abdel-Fatah, 2018; Wang et al., 2020).
59 The physicochemical characterization of this sample before and after treatment showed a
60 remarkable increase in removal rate from 1 hour to 3 hours of filtration; 54.8% after 1 hour and
61 73.28% after 3 hours for COD (73.28%) after 3 hours for BOD5 (26.48%) after 3 hours of
62 filtration for conductivity and a total elimination rate of turbidity that was achieved 99% after 3
63 hours of treatment(El Khalfaouy et al., 2017).
64 The combination of coagulation and adsorption processes for fulvic acid uptake shows that the
65 effectiveness of fulvic acid uptake is quite effective compared to the coagulation process. This
66 system provides sufficient contact time for PAC to adsorb organics compared to the system with
67 in-line adsorbent addition. More than 85% of FA was removed from water containing 8 mg. L⁻¹ of
68 FA.The removal efficiency also increased with the increase in mixing intensity and mixing time.
69 The permeate flux slightly improved when a membrane of pore size 0.22 μm was used with shorter
70 hydraulic residence time and lower PAC concentration so from this study it is evident that the
71 combined membrane process has a potential application for organic dye removal(Jirankova et al.,
72 2010).
73 The study on the effect of powder-activated carbon (PAC) adsorption on microfiltration (MF)
74 membrane performance are showed that PAC pretreatment offered high organic matter removal
75 rates for both dissolved organic carbon (DOC) and ultraviolet absorbance at 254 nm (UV254)
76 during 10–200 mg. L⁻¹ PAC dosage. The removal efficiencies of organic matter by MF membrane
77 filtration decreased with the increase of organic matter removal rate by PAC adsorption. PAC
78 mainly removed organic matter of about 3 kDa molecular weight (MW). MF membrane
79 maintained more than 5 kDa MW organic matter on the membrane after PAC adsorption. The
80 results of membrane filtration indicated that PAC pretreatment slightly promoted membrane flux,
81 regardless of PAC dosage. It seems that the organic matter fouling membrane was concentrated in
82 more than 3 kDa MW. PAC removed markedly less than 3 kDa MW organic matter and had less
83 effect on more than 3 kDa organic matter. Thus, PAC cannot reduce membrane fouling
84 Löwenberg, 2016).The AF2B/GAC process was able to reduce contaminants such as TSS, BOD,
85 COD, NH₃N and Chlorine by 98 %, 99 %, 97.3 %, 97.8 %, and 100 % respectively. Furthermore,
86 in the CNTs process, all pollutants are not detected until they meet drinking water quality
87 standards(Prayitno et al., 2022).

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89 **2. Materials and Methods**

90 **2.1 Materials and equipment preparation**

91 The feedwater used in this study is effluent from WWTP in the seaweed processing industry,
92 where the feedwater meets quality standards. Feed water contains odorless, color (15 - 25 Pt-Co),
93 Turbidity (8 - 14 NTU), pH (6 - 7), TSS (20 - 30 mg. L⁻¹), TDS (4,000 - 4,500 mg. L⁻¹), Total
94 hardness (700 - 900 mg. L⁻¹), BOD (30 - 40 mg. L⁻¹), and COD (60 - 80 mg. L⁻¹). Furthermore,
95 feedwater is fed into the influent column of 2,000 L to have flowed into the treatment system as
96 fed in the AMFIL process which operates continuously as Fig. 1.

97 The equipment used in this study includes an adsorption column, microfiltration column, pump, and
98 storage column. The adsorption column was made from PVC material with a diameter of 25 cm, a
99 height of 150 cm, and at the top of the column, there is a distributor for feedwater. The adsorption
100 column containing granular activated carbon (GAC) of 20 - 25 kg that GAC from YUJIA
101 production had specifications: total ash content max 4%, moisture content max 5%, apparent
102 density 0.4 - 0.5 g.cm⁻³, mean particle diameter 4 - 8 mesh, iodine number 900 mg. g⁻¹.

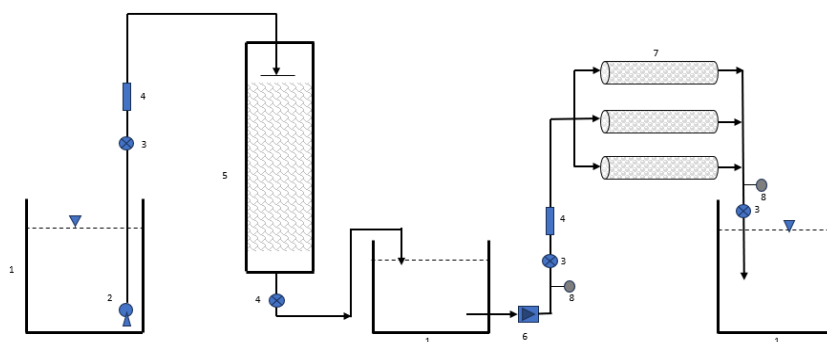
103 While the microfiltration membrane used cellulose membrane is type UF 280 with specifications:
104 PVC Alloy; Shell Material: U-PVC; Size: 8 x 40 inch; Filtration Method: Inside - Out; Flow Rate:
105 1500 - 2500 L.h⁻¹; Area: 28 m²; In/Out: 2"; pore diameter of 1 - 10 μm.

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107 **2.2 Operational procedure**

108 Real wastewater discharged from the WWTP in the seaweed processing industry was used as feed
 109 water in this study, where 2,000 L of feed water was collected in a storage column, and sampling
 110 was carried out to analyze the characteristics of the feed water. Furthermore, the feed water is
 111 flowed by a pump into an adsorption column that already contains 20 kg of granular activated
 112 carbon, while the effluent of the adsorption column is collected in a temporary storage column
 113 which is then sampled to analyze the effluent characteristics. The effluent from the adsorption
 114 column is pumped using a jet pump of 1-10 bar into the microfiltration column where the 2-
 115 membrane tube is arranged in parallel. The effluent from the membrane tube (filtrate), hereinafter
 116 known as process water, is sampled, and analyzed for its characteristics. The experiment was
 117 repeated using the weight of the adsorbent and the number of different membrane tubes.

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Notes:
 1. Holding Tank; 2. Submersible Pump; 3. Valve; 4. Flowmeter; 5. Adsorption Column;
 6. Jet Pump; 7. Membrane microfiltration; 8. Pressure Indicator

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Figure 1. The Process Flow of AMFIL System

125 **2.3 Analysis Method**

126 The samples were analyzed by using APHA standard procedures to investigate the pH, Ferro
 127 (Fe^{+2}), Mangan (Mn^{+2}), Turbidity, TDS, and Total Hardness. The pH was measured using a Fisher
 128 Scientific type pH meter (Model XCL25) previously calibrated with pH 6,7, and 8 standard
 129 solutions. Fe^{+2} and Mn^{+2} were measured using UV-Vis's spectrophotometry (Shimadzu UV-1280).
 130 Fe^{+2} analysis used the phenanthroline method, while Mn^{+2} analysis used the persulfate method.
 131 Fe^{+2} testing can be done using the phenanthroline complexing compound and analyzed at a
 132 wavelength of 480 nm – 560 nm. By making a standard curve, the concentration of Fe^{+2} in the
 133 sample can be determined. While determination of manganese in samples was carried out by the
 134 persulfate method at a wavelength of 400 - 700 nm according to the APHA standard method, with a
 135 calibration curve including standards of 0.5, 1.0, 1.5, 2.0 and 2.5 $mg \cdot L^{-1}$.

136 The determination of Total Dissolved Solids (TDS), according to the analytical method (MA115-
 137 S.S. 1.2), was carried out by filtering a sample portion (normally 100 ml) through a previously
 138 dried and weighed Whatman 934 AH glass microfiber filter (pore size 0.5 μm) under vacuum.
 139 When the filtration was completed, the residue was dried at 103-105 C overnight or at least 1 hour.
 140 The weight of dissolved solids is obtained by calculating the difference of the filter weights before
 141 and after drying. To determine pH using an electrical conductivity method in the form of
 142 measuring instrument pH Meter EZ-9910. The quality standard used as a reference is based on the
 143 Decree of the Minister of Health of the Republic of Indonesia No.32 of 2017.

144 Furthermore, to determine the adsorption capacity of total dissolved solids in the adsorption
 145 column, it is estimated to use equation 1.

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$$q_t = (C_o - C_t) (V/m) \dots\dots\dots (1)$$

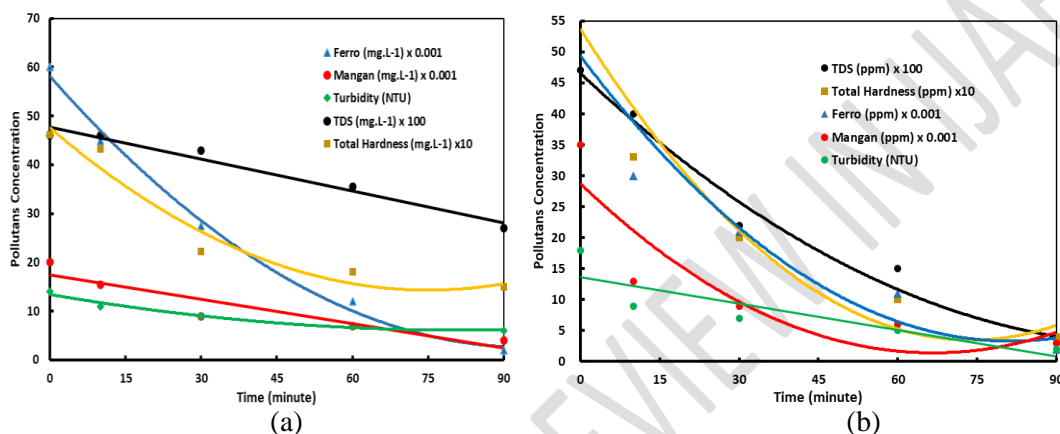
Which:

149 q_t is the adsorption capacity in mg. g^{-1} .
 150 C_o is the TDS initial concentration in mg. L^{-1}
 151 C_t is the TDS concentration in mg. L^{-1} attime.
 152 V is the solution volume in liter
 153 M is the adsorbent mass in gram

156 3. Results and Discussion

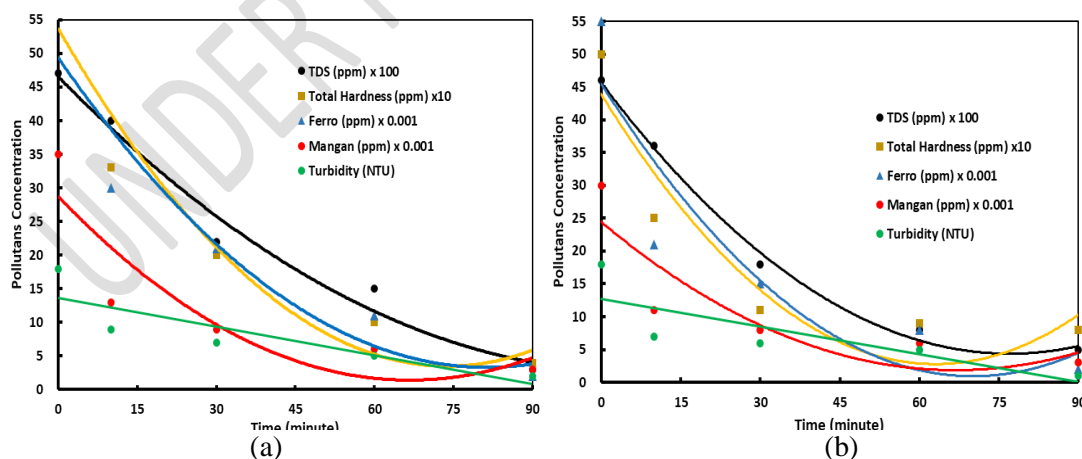
157 3.1 The effect of adsorbent weight

158 In the AMFIL process, especially in the adsorption column containing granular activated carbon,
 159 using an adsorbent weight of 20 kg and 25 kg and different feed characteristics, experimental
 160 results were obtained as shown in Figure 2.



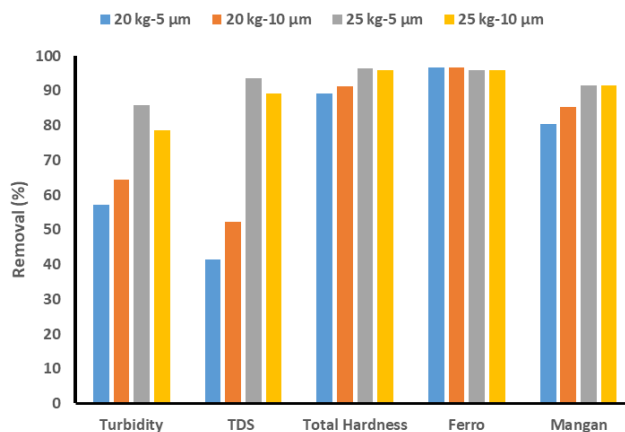
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 164 **Figure 2.** The performance of the adsorption column at different adsorbent weights
 165 a. Weight 20 kgb. Weight 25 kg

170 3.2 The effect of the size of the microfiltration membrane



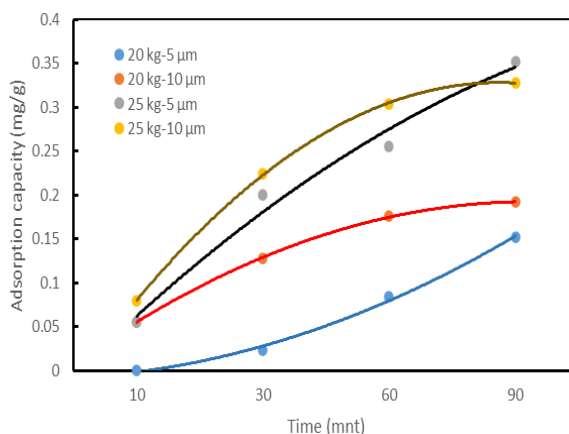
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 176 **Figure 3.** Performance of microfiltration membranes at different membrane sizes
 177 a. Size of membrane 5 μm b. Size of membrane 10 μm

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Figure 4. The performance of the AMFIL process



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Figure 5.

4. Conclusion

The AMFIL process with an adsorption capacity of 0.352 mg.g^{-1} and the size of a microfiltration membrane is $5\mu\text{m}$ are capable of processing effluent from WWTP into process water according to standard quality with the total removal pollutants such as turbidity, TDS, Fe, Mn, and pH are 86.67, 94.0, 95.83, 96.36, 93.33%, and (6.5 – 7.5), respectively.

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